

Crystal-field analyses for trivalent lanthanide ions in LiYF₄

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Abstract: Based on the completely parametric crystal-field model, the energy level parameters, including free-ion parameters and crystal-field parameters, obtained by fitting the experimental energy level data sets of Ln³⁺ in LiYF₄ were systematically analyzed. The results revealed that the regular variation trends of the major parameters at relatively low site symmetry still existed. The *g* factors of ground states were calculated using the parameters obtained from least-squares fitting. The results for Ce³⁺, Nd³⁺, Sm³⁺, Dy³⁺ and Yb³⁺ were in good agreement with experiment, while those of Er³⁺ deviated from experiment dramatically. Further study showed that the *g* factors depended strongly on *B*₄⁶, and a slightly different *B*₄⁶ value of -580 cm⁻¹ led to *g* factors agreeing well with the experimental values.

Keywords: crystal-field interactions; LiYF₄; lanthanide ions; variation trends; *g* factors; rare earths

The energy level parameters of trivalent lanthanide (Ln³⁺) ions in hosts can be determined by least-squares fitting to the experimental energy level data sets^[1-3]. Previous results on the energy level parameters of all Ln³⁺ in Cs₂NaLnCl₆^[4] with high site symmetry show that the major parameters vary smoothly across the Ln³⁺ series. If the trend holds for Ln³⁺ in all hosts, especially for those with low site symmetry, then it can be used to predict the crystal-field parameters (CFPs) for all Ln³⁺ from those for one particular Ln³⁺ ion in the same host. In particular, the CFPs of Ce³⁺ in crystals can be calculated from *ab-initio* calculations^[5-7].

Ln³⁺ doped luminescent materials are widely applied as phosphors, scintillators, laser materials and temperature sensors^[8-13]. Among these, Ln³⁺ doped LiYF₄ crystal has been deeply investigated with many spectroscopic and theoretical studies in the last several decades^[14-17]. Hence, we chose Ln³⁺ in LiYF₄ to study the variation trends of energy level parameters, especially for crystal-field (CF) interactions, for a low site symmetry case.

The LiYF₄ crystal has the scheelite structure. Ln³⁺ ion occupies Y³⁺ site at S₄ point symmetry, surrounded by a slightly distorted dodecahedron of eight F⁻ ions. As the distortion is very small, D_{2d} symmetry is chosen as a realistic approximation to the actual S₄ site symmetry^[18-20]. In D_{2d} symmetry, all the CFPs are real. Firstly, the energy level parameters of Ln³⁺ series in LiYF₄ were calculated from least-squares fitting to the experimental energy level data sets. Based on the fitted parameters, the *g*

factors of the ground states of Ln³⁺ were analyzed. The results for Ce³⁺, Nd³⁺, Sm³⁺, Dy³⁺ and Yb³⁺ were in good agreement with the experimental data, while those of Er³⁺ deviated from the experimental data. Further study showed that the *g* factors of Er³⁺ were sensitive to *B*₆⁶ value in a particular range, and a slightly different value of *B*₄⁶ = -580 cm⁻¹ could predict correct *g* factors.

1 Theoretical calculations

The energy levels of 4f^{*N*} configuration of Ln³⁺ in LiYF₄ were analyzed in terms of a completely parametric effective operator Hamiltonian:

$$H = E_{av} + \sum_k F^k f_k + \sum_i \zeta_i s_i \cdot l_i + \alpha L(L+1) + \beta G(G_2) + \gamma G(R_7) + \sum_h T^h t_h + \sum_s M^s m_s + \sum_k P^k p_k + H_{cf} \quad (1)$$

where all the parameters and operators have the same meaning as those in Ref. [21]. Specifically, *E*_{av} adjusts the configuration barycenter energy of the entire 4f^{*N*} configuration. *F*^{*k*} (*k*=2, 4, 6) are the Slater parameters and *f*^{*k*} represent the angular operator parts of the electrostatic interaction. *ζ*_{*i*} is the spin-orbit parameters and *s*_{*i*}·*l*_{*i*} represent the spin-orbit interactions. *α*, *β*, and *γ* are the parameters describing the two-body interactions. *L* is the total orbital angular momentum, and *G*(*G*₂) and *G*(*R*₇) are Casimir operators for the groups *G*₂ and *R*₇. For 4f^{*N*} and 4f¹⁴-*N* configurations of *N*≥3, the three-body parameters *T*^{*h*} (*h*=2, 3, 4, 6, 7, 8) and corresponding operators *t*_{*h*} are employed. *M*^{*s*} (*s*=0, 2, 4) are the Marvin inte-

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grals which describe the spin-spin and spin-other-orbit relativistic interactions between electrons. P^k ($k=2, 4, 6$) describe the two-body magnetic interactions. HCF is the CF Hamiltonian comprising the non-spherically symmetric crystal field.

In order to minimize the number of parameters in fitting the experimental data sets, M^2 and M^4 were constrained by the ratios $M^2=0.56M^0$, $M^4=0.38M^0$, and P^4 and P^6 were constrained by the ratios $P^4=0.75P^2$, $P^6=0.5P^2$ ^[22,23].

The CF Hamiltonian H_{CF} for D_{2d} symmetry can be written (in the formalism of Wybourne) as^[24]

$$H_{CF} = \sum_{k,q} B_q^k C_q^k = B_0^2 C_0^2 + B_0^4 C_0^4 + B_4^4 (C_4^4 + C_{-4}^4) + B_0^6 C_0^6 + B_4^6 (C_4^6 + C_{-4}^6) \quad (2)$$

where C_q^k is a spherical tensor of rank k , with components q ; B_q^k are the CFPs.

Based on the effective Hamiltonian, least-squares fitting to the experimental energy levels available was performed to obtain the energy level parameters across the Ln³⁺ series in LiYF₄ using the f-shell program package^[25]. Moreover, the magnetic properties were investigated. In the presence of an external magnetic field B , the Zeeman term can be written as^[26]

$$\hat{H}_z = \mu_B (g_s s + l) \cdot B \quad (3)$$

where μ_B is the Bohr magneton, $g_s=2.00232$ is the electron spin g factor, s and l are the spin and orbital angular momentum, respectively.

2 Results and discussion

The experimental energy level data sets of Ln³⁺ series in LiYF₄^[20,24,27-35] except Pm³⁺ and Gd³⁺ were systematically analyzed by least-squares fitting using the parametric effective Hamiltonian. The Gd³⁺ in LiYF₄ was not included due to lack of energy level data at low excited states^[36]. A summary of the calculated energy level parameters are presented in Table 1. Because of the limited amount of experimental data, some free-ion parameters were fixed using the values in Ref. [23]. The number of parameters (N_p) employed to fit the N_e energy levels is given in each case.

As can be seen, the Slater parameters F^k and the spin-orbit parameters ζ_{4f} have a regular increase across the Ln³⁺ series, which can be described by linear or second-order polynomial relation. The variation equations are as follows:

$$F^2=(63453\pm 1016)+(3137.7\pm 126.1)N \quad (4)$$

$$F^4=(46898\pm 757)+(1993.4\pm 94.0)N \quad (5)$$

$$F^6=(29965\pm 1099)+(1960.0\pm 136.4)N \quad (6)$$

$$\zeta_{4f}=(558.6\pm 13.4)+(83.03\pm 4.40)N+(7.443\pm 0.310)N^2 \quad (7)$$

The variation trends of CFPs of Ln³⁺ in LiYF₄ were not as regular as those in Cs₂NaLnF₆, but smooth variations were presented by linear fitting to CFPs across the Ln³⁺ series, as shown in Fig. 1(a). The CF strength parameter S is a quantitative measure of the overall CF interaction of Ln³⁺ within a particular host and is defined as^[37]

Table 1 Energy level parameters (in cm⁻¹) from fitting the 4f^N energy levels of Ln³⁺ series in LiYF₄^{a,b}

Ln ³⁺	Ce	Pr	Nd	Sm	Eu	Tb	Dy	Ho	Er	Tm	Yb
E_{avg}	1513	10202	24412	47597	63642	69519	55344	48159	35753	18004	4632
F^2		68955	72952	79515	82573	90972	90421	93512	97326	101938	
F^4		50505	52681	56766	59646	(64499)	(63928)	66084	67987	71553	
F^6		33098	35476	40078	43203	(45759)	(46657)	49765	53651	51359	
ζ_{4f}	630	748	877	1168	1329	1702	1895	2126	2377	2632	2916
α		23.3	21.0	[20.5]	21.6	[17.6]	[17.9]	[17.2]	18.1	[17.3]	
β		[-644]	-579	[-616]	-482	[-581]	[-628]	[-596]	-599	[-665]	
γ		[1413]	1446	[1565]	1140	[1792]	[1790]	[1839]	1870	[1936]	
T^2			210	[282]	[370]	[330]	[326]	[365]	380		
T^3			41	[26]	[40]	[40]	[23]	[37]	41		
T^4			74	[71]	[40]	[45]	[83]	[95]	69		
T^6			-293	[-257]	[-300]	[-365]	[-294]	[-274]	-356		
T^7			321	[314]	[380]	[320]	[403]	[331]	239		
T^8			205	[328]	[370]	[349]	[340]	[343]	390		
M^0		[1.88]	[1.85]	[2.38]	2.41	[2.70]	[4.46]	3.92	4.41	[4.93]	
P^2		[244]	304	[336]	332	[482]	[610]	[582]	795	[730]	
$B_0^2(\text{ff})$	354	512	391	370	339	413	360	386	325	339	446
$B_0^4(\text{ff})$	[-1043]	-1127	-1031	-757	-733	-867	-737	-629	-749	-627	-560
$B_0^6(\text{ff})$	[-1249]	-1239	-1271	-941	-1067	-1114	-943	-841	-1014	-913	-843
$B_4^4(\text{ff})$	[-65]	-85	-28	-67	-36	-41	-35	-33	-19	-39	[-23]
$B_4^6(\text{ff})$	[-1069]	-1205	-1046	-895	-764	-736	[-700]	-687	-635	-584	[-512]
σ	21.9	21.7	23.5	11.2	20.1	16.4	9.3	4.3	12.0	13.8	48.8
N_e	5	44	149	55	103	28	21	69	108	41	7
N_p	3	11	20	10	15	8	7	11	21	10	5

^a Values in brackets were fixed according to Ref. [23]; for Ce³⁺ and Yb³⁺, values were fixed according to variation trends; ^b Values in parentheses were constrained according to Ref. [23]; for Tb³⁺, $F^4/F^2=0.709$, $F^6/F^2=0.503$; for Dy³⁺, $F^4/F^2=0.707$, $F^6/F^2=0.516$

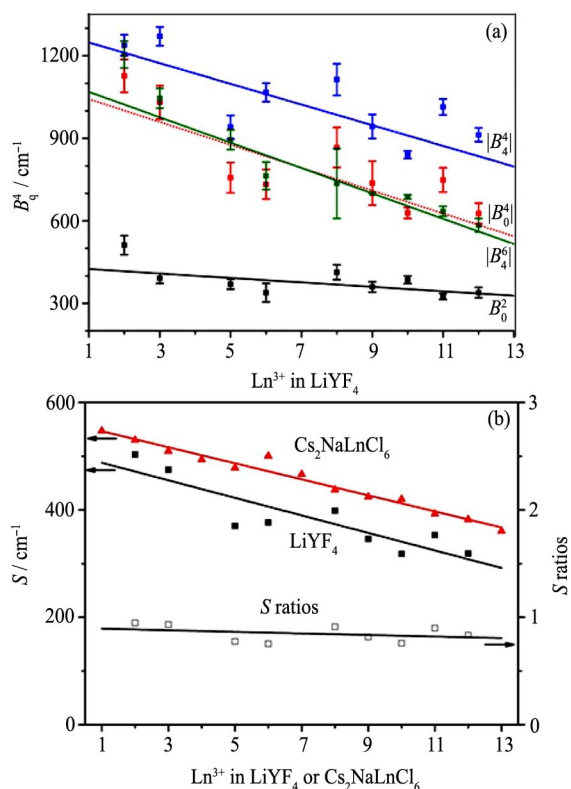


Fig. 1 (a) Variation trends of CFPs by fitting experimental energy levels of Ln^{3+} in LiYF_4 ; (b) S and S ratios of Ln^{3+} in LiYF_4 and $\text{Cs}_2\text{NaLnCl}_6$

$$S = \left(\frac{1}{3} \sum_k S_k^2 \right)^{1/2}, \quad S_k^2 = \frac{1}{2k+1} \sum_{q=-k}^k |B_q^k|^2 \quad (8)$$

Based on the calculated CFPs, CF strength parameters were analyzed, as shown in Fig. 1(b), together with S ratios of LiYF_4 and $\text{Cs}_2\text{NaLnCl}_6$. The S ratios in two hosts can be considered as a constant (0.909) across the Ln^{3+} series.

The equations of CFPs and CF strength parameters described by linear relation were also given as follows:

$$B_0^2 = (432.9 \pm 17.7) - (8.1 \pm 1.9)N \quad (9)$$

$$B_0^4 = (-1084.4 \pm 45.2) + (41.6 \pm 4.8)N \quad (10)$$

$$B_4^4 = (-1286.4 \pm 28.5) + (37.6 \pm 3.1)N \quad (11)$$

$$B_0^6 = (-68.4 \pm 14.0) + (3.5 \pm 1.7)N \quad (12)$$

$$B_4^6 = (-1115.2 \pm 27.6) + (46.1 \pm 3.1)N \quad (13)$$

$$S(\text{LYF}) = (504.1 \pm 25.8) - (16.3 \pm 3.2)N \quad (14)$$

$$S(\text{Cs}_2\text{NaLnCl}_6) = (561.5 \pm 6.3) - (14.9 \pm 0.8)N \quad (15)$$

$$S(\text{LYF})/S(\text{Cs}_2\text{NaLnCl}_6) = 0.901 \pm 0.063 \quad (16)$$

Note that the parameters of Ce^{3+} and Yb^{3+} were not included in the linear fitting of CFPs due to the limited experimental energy levels. The parameters of Ce^{3+} and Yb^{3+} were latter fitted using five and seven available experimental energy levels, respectively. Some parameters are poorly defined with the available experimental energy level data sets, and so are fixed to the values of variation trends during energy level fitting. Table 2 presents the linear variation trends of CFPs for Ln^{3+} in LiYF_4 .

Based on the calculated energy level parameters, some

Table 2 Linear variation trends of CFPs for Ln^{3+} in LiYF_4 (in cm^{-1})

Ln^{3+}	B_0^2 (ff)	B_0^4 (ff)	B_4^4 (ff)	B_0^6 (ff)	B_4^6 (ff)
Ce	425	-1043	-1249	-65	-1069
Pr	417	-1001	-1211	-61	-1023
Nd	409	-960	-1174	-58	-977
Pm	401	-918	-1136	-54	-931
Sm	392	-876	-1098	-51	-885
Eu	384	-835	-1061	-47	-839
Gd	376	-793	-1023	-44	-793
Tb	368	-752	-986	-40	-746
Dy	360	-710	-948	-37	-700
Ho	352	-668	-910	-33	-654
Er	344	-627	-873	-30	-608
Tm	336	-585	-835	-26	-562
Yb	328	-544	-798	-23	-516

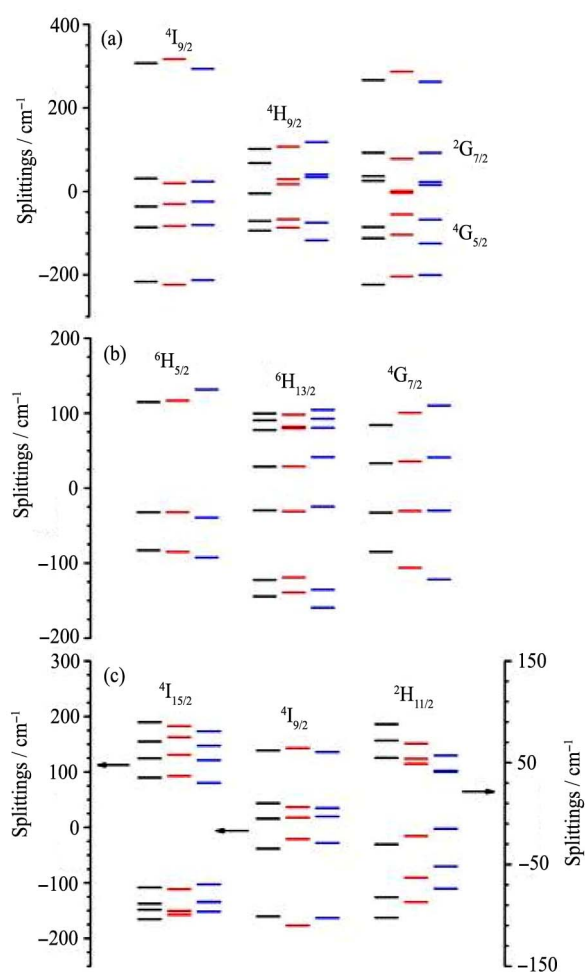


Fig. 2 Splittings of multiplets of Nd^{3+} (a), Sm^{3+} (b) and Er^{3+} (c) in LiYF_4 (The black bars (left), red bars (middle) and blue bars (right) denote the splittings from the experimental data, the fitted data and the calculated data using the trend values, respectively)

multiplets of Nd^{3+} , Sm^{3+} and Er^{3+} were analyzed to compare with the experimental data, as presented in Fig. 2. It shows that the splittings using the two sets of energy level parameters are both consistent with the corresponding experimental values.

The values of CFPs for Ce³⁺ in LiYF₄ from *ab-initio* calculation^[7] are $B_0^2=310\text{ cm}^{-1}$, $B_0^4=-1104\text{ cm}^{-1}$, $B_4^4=-1418\text{ cm}^{-1}$, $B_0^6=-70\text{ cm}^{-1}$, $B_4^6=(-1140+237i)\text{ cm}^{-1}$, and the CF strength parameter is 520 cm^{-1} , merely 3.4% larger than that obtained using the CFPs from least-squares fitting. This indicates that, in a given host, the CFPs from *ab-initio* calculation for Ce³⁺ can be applied to predict the corresponding parameters and then the complicated energy level structures of other Ln³⁺ ions, where *ab-initio* calculations are computationally formidable.

Table 3 *g* factors of the ground states of Ln³⁺ in LiYF₄

Ln ³⁺	Ground states	<i>g_z</i>			<i>g_x</i>		
		Expt. ^a	Calc. ^b	Calc. ^c	Expt. ^a	Calc. ^b	Calc. ^c
Ce ³⁺	² F _{5/2}	2.765	2.761	2.890	1.473	1.576	1.532
Nd ³⁺	⁴ I _{9/2}	1.987	1.550	1.488	2.554	2.674	2.671
Sm ³⁺	⁶ H _{5/2}	0.410	0.455	0.450	0.644	0.739	0.727
Dy ³⁺	⁶ H _{15/2}	1.112	0.899	0.568	9.219	9.168	9.135
Er ³⁺	⁴ I _{15/2}	3.137	8.452	4.974	8.105	4.376	7.524
Yb ³⁺	² F _{7/2}	1.331	1.338	1.310	3.917	3.978	3.980

^a Experimental values from Refs. [20, 38, 39]; ^{b,c} Values calculated using the sets of Slater parameters, spin-orbit parameters and CFPs from least-squares fitting and variation trends, respectively

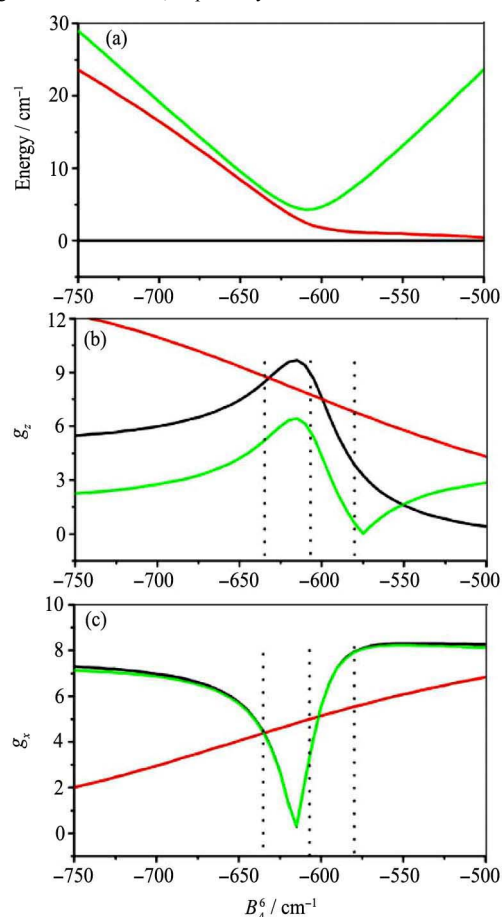


Fig. 3 Variation curves of the position of the lowest three Kramer's doublets (black, red and green lines) (a) and corresponding *g_z* (b) and *g_x* (c) of Er³⁺ with B_4^6 varying from -800 to -500 cm^{-1} (The values of B_4^6 marked at -635 , -608 and -580 cm^{-1} (dotted vertical bars) are from energy level fitting, variation trends and a different value for *g* factors)

The magnetic properties were then investigated using the calculated parameters from both least-squares fitting and variation trends. Table 3 presents the calculated *g* factors of the ground states of Ln³⁺ in LiYF₄. The results for Ce³⁺, Nd³⁺, Sm³⁺, Dy³⁺ and Yb³⁺ are in good agreement with experimental values, while those for Er³⁺ deviate from the experimental data. We found that the energies of the first few levels and the *g* factors are sensitive to the value of B_4^6 in a particular range, but not so sensitive to other CFPs. Fig. 3 presents the variations of the position of the lowest three Kramer's doublets and the corresponding *g_z* and *g_x* of Er³⁺ with B_4^6 value varying from -800 to -500 cm^{-1} . The *g* factors are closer to the experimental values when the strength of B_4^6 decreases, in accord with the variation trends. Optimized values $B_4^6=-580\text{ cm}^{-1}$ and $B_0^2=344\text{ cm}^{-1}$ predict $g_z=3.062$ and $g_x=8.037$, agreeing well with experiment.

3 Conclusions

The energy level parameters were analyzed by fitting the experimental energy level data sets across the Ln³⁺ series in LiYF₄. The linear or second-order polynomial fitting of the parameters to the number of 4f electrons were performed. To some extent, the regular variation trends of the major parameters for a low symmetry site were maintained. Since CFPs of Ce³⁺ in a given host could be calculated from *ab-initio* calculation, the CFPs of other Ln³⁺ ions could be obtained by following the trends, so that the energy levels of all Ln³⁺ ions in the same host could be calculated. Based on the CF analyses, the *g* factors of the ground states of Ln³⁺ in LiYF₄ were also analyzed. The results for Ce³⁺, Nd³⁺, Sm³⁺, Dy³⁺ and Yb³⁺ were in good agreement with the experimental data, while those of Er³⁺ deviated from the experimental data. Further study showed that the *g* factors of Er³⁺ were sensitive to B_4^6 value, and a slightly different value of $B_4^6=-580\text{ cm}^{-1}$ could predict correct *g* factors. This suggests that *g* factors can be taken into account to check or to increase the reliability of CFPs.

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